

Table 2.14. Lifetimes, radiative efficiencies and direct (except for CH₄) GWP's relative to CO₂. For ozone-depleting substances and their replacements, data are taken from IPCC/TEAP (2005) unless otherwise indicated.

Industrial Designation or Common Name (years)	Chemical Formula	Lifetime (years)	Radiative Efficiency (W m ⁻² ppb ⁻¹)	Global Warming Potential for Given Time Horizon			
				SARF (100-yr)	20-yr	100-yr	500-yr
Carbon dioxide	CO ₂	See below ^a	1.4x10 ⁻⁵	1	1	1	1
Methane ^c	CH ₄	12 ^c	3.7x10 ⁻⁴	21	72	25	7.6
Nitrous oxide	N ₂ O	114	3.03x10 ⁻³	310	289	298	153
Substances controlled by the Montreal Protocol							
CFC-11	CCl ₃ F	45	0.25	3,800	6,730	4,750	1,620
CFC-12	CCl ₂ F ₂	100	0.32	8,100	11,000	10,900	5,200
CFC-13	CClF ₃	640	0.25		10,800	14,400	16,400
CFC-113	CCl ₂ FCF ₂	85	0.3	4,800	6,540	6,130	2,700
CFC-114	CClF ₂ CClF ₂	300	0.31		8,040	10,000	8,730
CFC-115	CClF ₂ CF ₃	1,700	0.18		5,310	7,370	9,990
Halon-1301	CBrF ₃	65	0.32	5,400	8,480	7,140	2,760
Halon-1211	CBrClF ₂	16	0.3		4,750	1,890	575
Halon-2402	CBrF ₂ CBrF ₂	20	0.33		3,680	1,640	503
Carbon tetrachloride	CCl ₄	26	0.13	1,400	2,700	1,400	435
Methyl bromide	CH ₃ Br	0.7	0.01		17	5	1
Methyl chloroform	CH ₃ CCl ₃	5	0.06		506	146	45
HCFC-22	CHClF ₂	12	0.2	1,500	5,160	1,810	549
HCFC-123	CHCl ₂ CF ₃	1.3	0.14	90	273	77	24
HCFC-124	CHClFCF ₃	5.8	0.22	470	2,070	609	185
HCFC-141b	CH ₃ CCl ₂ F	9.3	0.14		2,250	725	220
HCFC-142b	CH ₃ CClF ₂	17.9	0.2	1,800	5,490	2,310	705
HCFC-225ca	CHCl ₂ CF ₂ CF ₃	1.9	0.2		429	122	37
HCFC-225cb	CHClFCF ₂ CClF ₂	5.8	0.32		2,030	595	181
Hydrofluorocarbons							
HFC-23	CHF ₃	270	0.19	11,700	12,000	14,800	12,200
HFC-32	CH ₂ F ₂	4.9	0.11	650	2,330	675	205
HFC-125	CHF ₂ CF ₃	29	0.23	2,800	6,350	3,500	1,100
HFC-134a	CH ₂ FCF ₃	14	0.16	1,300	3,830	1,430	435
HFC-143a	CH ₃ CF ₃	52	0.13	3,800	5,890	4,470	1,590
HFC-152a	CH ₃ CHF ₂	1.4	0.09	140	437	124	38
HFC-227ea	CF ₃ CHFCF ₃	34.2	0.26	2,900	5,310	3,220	1,040
HFC-236fa	CF ₃ CH ₂ CF ₃	240	0.28	6,300	8,100	9,810	7,660
HFC-245fa	CHF ₂ CH ₂ CF ₃	7.6	0.28		3,380	1030	314
HFC-365mfc	CH ₃ CF ₂ CH ₂ CF ₃	8.6	0.21		2,520	794	241
HFC-43-10mee	CF ₃ CHFCFCF ₂ CF ₃	15.9	0.4	1,300	4,140	1,640	500
Perfluorinated compounds							
Sulphur hexafluoride	SF ₆	3,200	0.52	23,900	16,300	22,800	32,600
Nitrogen trifluoride	NF ₃	740	0.21		12,300	17,200	20,700
PFC-14	CF ₄	50,000	0.10	6,500	5,210	7,390	11,200
PFC-116	C ₂ F ₆	10,000	0.26	9,200	8,630	12,200	18,200

Table 2.14 (continued)

Industrial Designation or Common Name (years)	Chemical Formula	Lifetime (years)	Radiative Efficiency (W m ⁻² ppb ⁻¹)	Global Warming Potential for Given Time Horizon			
				SARF (100-yr)	20-yr	100-yr	500-yr
Perfluorinated compounds (continued)							
PFC-218	C ₃ F ₈	2,600	0.26	7,000	6,310	8,830	12,500
PFC-318	c-C ₄ F ₈	3,200	0.32	8,700	7,310	10,300	14,700
PFC-3-1-10	C ₄ F ₁₀	2,600	0.33	7,000	6,330	8,860	12,500
PFC-4-1-12	C ₅ F ₁₂	4,100	0.41		6,510	9,160	13,300
PFC-5-1-14	C ₆ F ₁₄	3,200	0.49	7,400	6,600	9,300	13,300
PFC-9-1-18	C ₁₀ F ₁₈	>1,000 ^d	0.56		>5,500	>7,500	>9,500
trifluoromethyl sulphur pentafluoride	SF ₅ CF ₃	800	0.57		13,200	17,700	21,200
Fluorinated ethers							
HFE-125	CHF ₂ OCF ₃	136	0.44		13,800	14,900	8,490
HFE-134	CHF ₂ OCHF ₂	26	0.45		12,200	6,320	1,960
HFE-143a	CH ₃ OCF ₃	4.3	0.27		2,630	756	230
HCFE-235da2	CHF ₂ OCHClCF ₃	2.6	0.38		1,230	350	106
HFE-245cb2	CH ₃ OCF ₂ CHF ₂	5.1	0.32		2,440	708	215
HFE-245fa2	CHF ₂ OCH ₂ CF ₃	4.9	0.31		2,280	659	200
HFE-254cb2	CH ₃ OCF ₂ CHF ₂	2.6	0.28		1,260	359	109
HFE-347mcc3	CH ₃ OCF ₂ CF ₂ CF ₃	5.2	0.34		1,980	575	175
HFE-347pcf2	CHF ₂ CF ₂ OCH ₂ CF ₃	7.1	0.25		1,900	580	175
HFE-356pcc3	CH ₃ OCF ₂ CF ₂ CHF ₂	0.33	0.93		386	110	33
HFE-449sl (HFE-7100)	C ₄ F ₉ OCH ₃	3.8	0.31		1,040	297	90
HFE-569sf2 (HFE-7200)	C ₄ F ₉ OC ₂ H ₅	0.77	0.3		207	59	18
HFE-43-10pccc124 (H-Galden 1040x)	CHF ₂ OCF ₂ OC ₂ F ₄ OCHF ₂	6.3	1.37		6,320	1,870	569
HFE-236ca12 (HG-10)	CHF ₂ OCF ₂ OCHF ₂	12.1	0.66		8,000	2,800	860
HFE-338pcc13 (HG-01)	CHF ₂ OCF ₂ CF ₂ OCHF ₂	6.2	0.87		5,100	1,500	460
Perfluoropolyethers							
PFPME	CF ₃ OCF(CF ₃)CF ₂ OCF ₂ OCF ₃	800	0.65		7,620	10,300	12,400
Hydrocarbons and other compounds – Direct Effects							
Dimethylether	CH ₃ OCH ₃	0.015	0.02		1	1	<<1
Methylene chloride	CH ₂ Cl ₂	0.38	0.03		31	8.7	2.7
Methyl chloride	CH ₃ Cl	1.0	0.01		45	13	4

Notes:

- ^a The CO₂ response function used in this report is based on the revised version of the Bern Carbon cycle model used in Chapter 10 of this report (Bern2.5CC; Joos et al. 2001) using a background CO₂ concentration value of 378 ppm. The decay of a pulse of CO₂ with time *t* is given by

$$a_0 + \sum_{i=1}^3 a_i \cdot e^{-t/\tau_i}$$

Where $a_0 = 0.217$, $a_1 = 0.259$, $a_2 = 0.338$, $a_3 = 0.186$, $\tau_1 = 172.9$ years, $\tau_2 = 18.51$ years, and $\tau_3 = 1.186$ years.

- ^b The radiative efficiency of CO₂ is calculated using the IPCC (1990) simplified expression as revised in the TAR, with an updated background concentration value of 378 ppm and a perturbation of +1 ppm (see Section 2.10.2).
- ^c The perturbation lifetime for methane is 12 years as in the TAR (see also Section 7.4). The GWP for methane includes indirect effects from enhancements of ozone and stratospheric water vapour (see Section 2.10.3.1).
- ^d Shine et al. (2005c), updated by the revised AGWP for CO₂. The assumed lifetime of 1,000 years is a lower limit.
- ^e Hurley et al. (2005)
- ^f Robson et al. (2006)
- ^g Young et al. (2006)